

N,N*-Dibenzyl-4-(4,4-diphenylbuta-1,3-dienyl)-3-methylaniline*An-Shu Wu, Xiang-Gao Li,*
De-Shun Xu and Shi-Rong Wang**School of Chemical Engineering and
Technology, Tianjin University, Tianjin 300072,
People's Republic of China

Correspondence e-mail: wuanshujxys@126.com

The title compound, C₃₇H₃₃N, was synthesized by the Wittig–Horner reaction of 4-(*N,N*-dibenzylamino)-2-methylbenzaldehyde and the phosphonate carbanion, derived from 1,1-diphenyl-3-chloropropylene and triethyl phosphite by the Arbuzov reaction. The butadiene fragment has a planar *cisoid* conformation, while the two benzyl fragments are almost parallel to each other, forming a dihedral angle of 9.8 (2)°.

Received 7 November 2005
Accepted 14 November 2005
Online 19 November 2005**Key indicators**Single-crystal X-ray study
T = 294 K
Mean σ (C–C) = 0.004 Å
R factor = 0.054
wR factor = 0.131
Data-to-parameter ratio = 14.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**Comment**

Hole transporting materials (HTMs) play an important role in the fabrication of organic photo-conductors (OPCs) (Wu *et al.*, 2005) and organic light-emitting diodes (OLEDs) (Li *et al.*, 2005; Satoh *et al.*, 2003). Substances containing the butadiene structure have been widely investigated, because of their easy preparation by the Wittig reaction and favorable hole-transport properties when used as HTMs (Enokida & Hirohashi, 1991). We report here the synthesis and crystal structure of the title compound, (I), synthesized by the Wittig–Horner reaction of 2-methyl-4-(*N,N*-dibenzyl)aminobenzaldehyde and the phosphonate carbanion, derived from 1,1-diphenyl-3-chloropropylene and triethyl phosphite by the Arbuzov reaction.

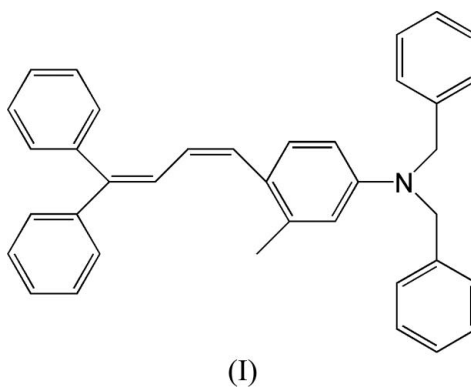


Fig. 1 shows the molecular structure of (I). The butadiene fragment C13=C14–C15=C16 is planar, with a maximum deviation of 0.003 (7) Å for atom C15, and has a *cisoid* conformation. Both the C1/C13/C7 plane and the plane of the C17–C22 benzene ring show substantial deviations from the butadiene plane, forming dihedral angles with the latter of 9.3 (2)° and 130.3 (2)°, respectively. The dihedral angles formed by the plane of the butadiene fragment with the planes of the C1–C6 and C7–C12 rings are 69.6 (2)° and 39.2 (2)°, respectively. The two benzyl fragments (C24–C30) and (C31–C37) are almost parallel to each other, forming a dihedral angle of 9.8 (2)°. The C1–C13 [1.487 (3) Å] and C7–C13

[1.480 (3) Å] bond distances are a little longer than C16–C17 [1.467 (3) Å].

Experimental

A mixture of 1,1-diphenyl-3-chloropropylene (11.4 g, 0.05 mol) and triethyl phosphite (18.0 ml, 0.05 mol) was refluxed in 50 ml of xylene for 10 h and the xylene was then removed *in vacuo*. After cooling, 2-methyl-4-(*N,N*-dibenzyl)aminobenzaldehyde (12.6 g, 0.04 mol) and 100 ml of dimethylformamide were added to the flask containing the residue. Thereafter, 4.5 g (0.04 mol) of potassium *tert*-butoxide was added in small portions. The resulting mixture was stirred for 5 h and then poured into methanol. The resulting precipitate was separated from the liquid by filtration, purified by silica gel column chromatography (eluent: toluene/ethyl acetate = 2:1), recrystallized from hexane, and dried to obtain yellow crystals (yield: 41.2%, m.p. 420 K).

Crystal data

$C_{37}H_{33}N$	$Z = 2$
$M_r = 491.64$	$D_x = 1.172 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 6.1629 (16) \text{ \AA}$	Cell parameters from 1753 reflections
$b = 15.935 (4) \text{ \AA}$	$\theta = 2.5\text{--}26.3^\circ$
$c = 16.323 (4) \text{ \AA}$	$\mu = 0.07 \text{ mm}^{-1}$
$\alpha = 117.824 (4)^\circ$	$T = 294 (2) \text{ K}$
$\beta = 90.749 (5)^\circ$	Plate, yellow
$\gamma = 99.054 (5)^\circ$	$0.40 \times 0.16 \times 0.08 \text{ mm}$
$V = 1393.2 (6) \text{ \AA}^3$	

Data collection

Bruker SMART 1000 CCD area-detector diffractometer	4873 independent reflections
φ and ω scans	2678 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Bruker, 1997)	$R_{\text{int}} = 0.027$
$T_{\text{min}} = 0.974$, $T_{\text{max}} = 0.995$	$\theta_{\text{max}} = 25.0^\circ$
7091 measured reflections	$h = -7 \rightarrow 5$
	$k = -18 \rightarrow 18$
	$l = -15 \rightarrow 19$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0464P)^2 + 0.1576P]$
$R[F^2 > 2\sigma(F^2)] = 0.054$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.131$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.14 \text{ e \AA}^{-3}$
4873 reflections	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$
344 parameters	
H-atom parameters constrained	

H atoms were positioned geometrically and refined in the riding-model approximation, with C–H = 0.93–0.98 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic and methine H atoms [$U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for CH_3].

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics:

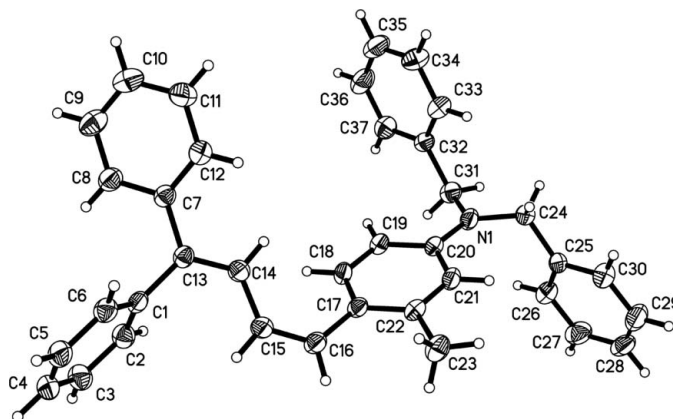


Figure 1

The molecular structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small circles of arbitrary radii.

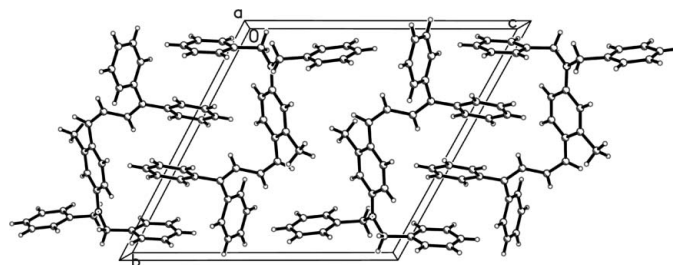


Figure 2

Packing diagram of (I).

SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

This work was financed by the National High-Technology Research and Development Program of China (grant No. 2002AA325050).

References

- Bruker (1997). SADABS, SMART, SAINT and SHELXTL. Versions 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Enokida, T. & Hirohashi, R. (1991). *J. Appl. Phys.* **70**, 6908–6912.
- Li, J. Y., Li, Y. Q., Lee, C. S., Kwong, H. L. & Lee, S. (2005). *Chem. Mater.* **17**, 1208–1212.
- Satoh, N., Cho, J. S., Higuchi, M. & Yamamoto, K. (2003). *J. Am. Chem. Soc.* **125**, 8104–8105.
- Sheldrick G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Wu, A.-S., Li, X.-G., Wang, S.-R. & Xue, J.-Q. (2005). *Gongneng Cailiao*, **36**, 708–710. (In Chinese.)